Q.P. Code: 16HS604

Reg. No:

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR

(AUTONOMOUS)

B.Tech I Year II Semester Supplementary Examinations July-2021 ENGINEERING CHEMISTRY

(Common to AGE, CE, EEE & ME)

Time: 3 hours Max. Marks: 60

(Answer all Five Units $5 \times 12 = 60$ Marks)

UNIT-I

1 Define Corrosion. Discuss various factors influencing the rate of corrosion. 12M

a Write a note on sacrificial anodic protection? 2 6M

b Discuss Hydrogen - Oxygen fuel cell.

6M

UNIT-II

What is the principle of EDTA method? Describe the estimation of hardness by 12M EDTA method.

OR

a Explain scale and Sludge formation in boilers. How are they removed? 6M

b What is Priming and Foaming?

6M

UNIT-III

Define Lubricant. Discuss the important functions of Lubricants. Explain the 12M mechanism of Lubrication.

OR

a Explain the manufacture of power alcohol. **6M**

b Explain Bergius process for the manufacture of synthetic petrol.

6M

UNIT-IV

a Explain different types of Polymerization with suitable examples.

6M

b Write the preparation, properties and uses of Buna-S rubber and Nitrile rubber?

6M

What are conducting polymers? How are they syntheized? Write important 12M engineering applications?

OR

UNIT-V

Define Cement. Explain the manufacture of Portland Cement.

12M

OR

10 a What are the characteristics of a good refractory?

4M

b Explain the properties of thermal spalling, porosity, dimensional stability and **8M** thermal conductivity of the refractories.

*** END ***